

**Zeitschrift:** Schweizerische mineralogische und petrographische Mitteilungen =  
Bulletin suisse de minéralogie et pétrographie

**Band:** 78 (1998)

**Heft:** 3

**Artikel:** Ordovician Barrow-type metamorphism in the Strona-Ceneri Zone  
(Northern Italy) dated by U-Pb on staurolite

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**DOI:** <https://doi.org/10.5169/seals-59296>

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# Ordovician Barrow-type metamorphism in the Strona-Ceneri Zone (Northern Italy) dated by U–Pb on staurolite

by Rolf L. Romer<sup>1</sup> and Leander Franz<sup>1,2</sup>

## Abstract

The crustal segment of the Strona-Ceneri Zone (SCZ) in northern Italy shows a complex polymetamorphic evolution that is bracketed by the emplacement age of 450–460 Ma old granitoids and the Middle Carboniferous age of mica and amphibole presumably related with the D3-phase. The age of the strongest tectonometamorphic event (D2) in the SCZ, which took place under amphibolite facies conditions at  $600 \pm 30$  °C and  $8 \pm 1$  kbar, resulted in the growth of abundant syntectonic staurolite in a garnet mica schist formed during the D2 clockwise prograde P-T evolution that is typical for Barrow-type metamorphism. The U–Pb and Pb–Pb systematics of staurolite demonstrate that staurolite formed during an Ordovician event and had lost lead during a Variscan overprint. Lead loss occurred preferably from microfractures and crystal defects, whereas lead from structural sites seems not to have been affected. The staurolite gives an Ordovician age for the D2 event, which implies that the metamorphic and structural memory of the SCZ was acquired during at least two distinct tectonometamorphic events.

*Keywords:* Strona-Ceneri Zone, Southern Alps, U–Pb, staurolite, Ordovician, tectonometamorphism.

## Introduction

Radiometric age determination in polymetamorphic terranes is often problematic. (1) Minerals with high closing temperatures such as monazite and zircon commonly cannot be related uniquely to a mineral assemblage distinctive for certain P-T conditions. Therefore, the age obtained from such accessory minerals cannot always be related uniquely with one of the tectonometamorphic events reflected in the mineral assemblage. (2) Minerals such as amphibole and mica, which are easily related to critical mineral assemblages, have closing temperatures for the Rb–Sr, K–Ar, and <sup>39</sup>Ar–<sup>40</sup>Ar systems, depending on the obtained metamorphic grade of the rock, that may be lower than the crystallization temperature of the mineral. The ages obtained from these minerals may represent minimum ages for this event rather than true ages.

These limitations of geochronometers leave space for conflicting interpretations: The crustal segment of the Strona-Ceneri Zone (SCZ) in northern Italy shows a complex polymetamorphic

evolution bracketed by the emplacement age of 450–460 Ma old granitoids and Middle Carboniferous ages for mica and amphibole presumably related with the D3-phase (ZURBRIGGEN et al., 1997). The age of the strongest tectonometamorphic event in the SCZ (D2 in ZURBRIGGEN et al., 1997; D1 in BORIANI and VILLA, 1997), which took place under amphibolite facies conditions of  $600 \pm 30$  °C and  $8 \pm 1$  kbar, is debated. ZURBRIGGEN et al. (1997) interpret the Barrow-type metamorphism associated with the D2 event as an Ordovician event. The cooling ages on mica and amphibole as well as Pb–Pb ages on garnet of D3 deformed dikes (ZURBRIGGEN et al., 1998) demonstrate a second tectonometamorphism in mid-Carboniferous. This implies that the structural and metamorphic scenario in the Strona-Ceneri Zone was acquired during at least two unrelated orogenic events. In contrast, BORIANI and VILLA (1997) related the entire structural and metamorphic memory of rocks in the SCZ to one single Variscan tectonothermal event. The problem in part arises from the contrasting interpretation of radiometric ages. ZURBRIGGEN et al. (1997) inter-

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pret the monazite U–Pb ages as the time of metamorphism and the K–Ar and  $^{39}\text{Ar}$ – $^{40}\text{Ar}$  ages as minimum ages after a second metamorphism. In contrast, BORIANI and VILLA (1997) interpret the K–Ar and  $^{39}\text{Ar}$ – $^{40}\text{Ar}$  ages as crystallization ages and refer to the older U–Pb ages as irrelevant to the metamorphism.

Controversies of this kind can be resolved or could be avoided if the dated minerals (1) are uniquely related to the structures of interest and (2) are dated by a method that remains a closed system to temperatures significantly higher than the thermal peak experienced by the dated rock. For the SCZ, staurolite represents such a mineral. Staurolite forms (1) a major constituent of S2 structures (as used by ZURBRIGGEN et al., 1997), (2) has a blocking temperature for the U–Pb system that is higher than the thermal maximum experienced by its host rock (see DAHL, 1997), and (3) has no inherited components. We demonstrate that staurolite yields an Ordovician age, which supports claims that the metamorphic and structural memory of the SCZ was acquired during at least two unrelated orogenic events and refutes

models that rule out a pre-Variscan structural and metamorphic history of the SCZ.

### Relation between Ivrea-Verbano Zone (IVZ) and Strona-Ceneri Zone (SCZ)

The Ivrea Crustal Section is located in northern Italy and forms a part of the pre-Alpine basement of the Southern Alps (Fig. 1). The section, which has a length of about 150 km and a width of up to 60 km, is bound to the northwest by the Insubric Line. In the south, the basement rocks of the Ivrea Crustal Segment are overlain by Permo-Carboniferous sedimentary and volcanic rocks and Mesozoic shelf sequences. The segment is subdivided in the Ivrea-Verbano Zone (IVZ) to the northwest and the Strona-Ceneri Zone (SCZ) to the southeast. IVZ and SCZ are separated by the Pogallo Line, which has been interpreted as a major tilted normal fault zone (HODGES and FOUNTAIN, 1984; HANDY, 1987) and the Cossato-Mergozzo-Brissago (CMBL) Line (BORIANI et al., 1990). The protoliths of the IVZ metamorphic

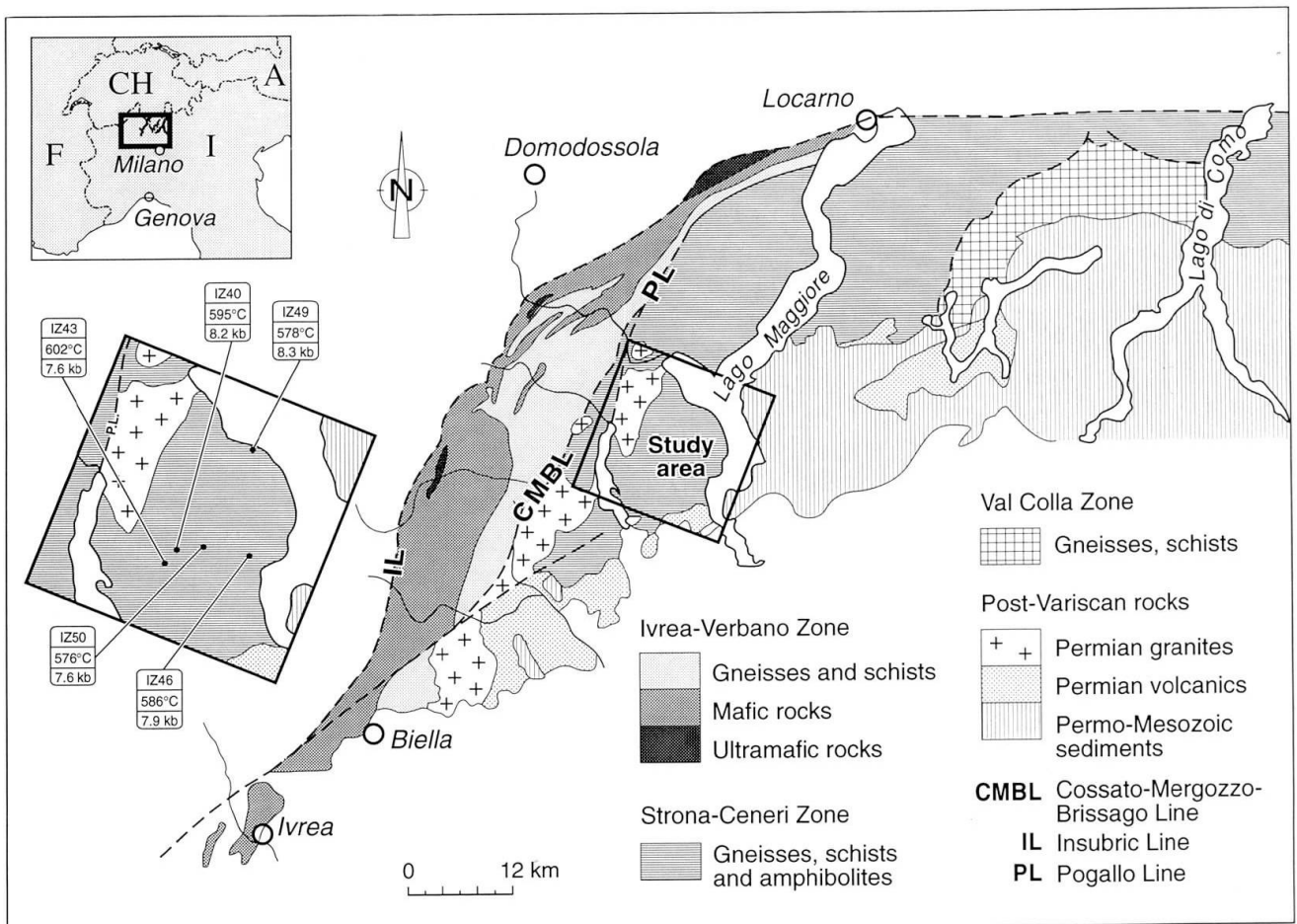


Fig. 1 Geographic position of the Ivrea crustal segment with sample location (centered) and peak metamorphic conditions of the selected mica schists.

rocks were probably deposited/erupted in an accretionary wedge (HANDY et al., in press). The only age estimate for the volcanism and sedimentation in the IVZ originates from SHRIMP analyses on 350 Ma old cores in zircon (VAVRA et al., 1996, see also U–Pb zircon data of KÖPPEL and GRÜNENFELDER [1971, 1978]). The IVZ experienced intense Permo-Carboniferous underplating by mafic and intermediate intrusive rocks at the mantle-crust boundary (HANDY and ZINGG, 1991). Strong thermal effects related with the magmatic underplating and the penetrative extensional deformation in the overlying metamorphic rocks led to the re-equilibration of most geothermobarometers and geochronometers (HENK et al., 1997). Relics of the pre-underplating evolution of the IVZ are limited to the southeastern border of the zone and include kyanite overgrown by sillimanite in paragneiss (BERTOLANI, 1959; CAPEDE, 1971; BORIANI and SACCHI, 1973; HANDY, 1986) and rare, strongly altered coronas around garnet (BORIANI and PEYRONEL PAGLIANI, 1968). These relics indicate an early, pre- to syn-accretionary phase of subduction and pressure-dominated metamorphism in the IVZ (see HANDY et al., in press).

Similar to the IVZ, the SCZ was affected by a multistage tectonometamorphic evolution. The earliest records of metamorphism in the SCZ, however, are distinctly older than in the IVZ. The protoliths of the Strona-Ceneri sediments probably were part of a Late-Proterozoic to Early Paleozoic subduction accretion complex that was intruded by Ordovician granitoids (ZURBRIGGEN et al., 1997). Xenoliths within these granitoids include relics of MOR-basalts that experienced eclogite-facies metamorphism and paragneisses that underwent D1 deformation under amphibolite-facies conditions. These eclogite and D1-metasediment xenoliths show a penetrative D2 deformation under prograde amphibolite-facies conditions. Textural criteria in the orthogneisses and the adjacent metasediments demonstrate that these granitoids intruded during D2 (ZURBRIGGEN et al., 1997). Therefore, the age of deformation D2 is given by the U–Pb age of zircon and monazite of e.g. the Giumello gneiss dated at  $457 \pm 4$  Ma (discordia intercept, RAGETTLI, 1993) and  $450 \pm 10$  Ma (KÖPPEL and GRÜNENFELDER, 1971) and the Ceneri gneiss dated at  $456 \pm 4$  Ma (concordant, RAGETTLI, 1993). P–T conditions during this event are estimated at 570–630 °C and 7–9 kbar (FRANZ et al., 1996; ZURBRIGGEN et al., 1997). The SCZ was overprinted by large scale D3 folds at ca 330 Ma (ZURBRIGGEN et al., 1998), so called "Schlingen", that formed under lower amphibolite-facies conditions (500–550 °C, 4–5 kbar; see FRANZ et al., 1996; HANDY et al., in press). The

D3-event, which may be related to the tectonic underplating/accretion of the IVZ with the SCZ, was accompanied by dike-magmatism and wiped out the geochronological features of the preceding D2 event in wide parts of the SCZ (HANDY et al., in press). The border zone of the SCZ was additionally affected by the thermal effects of the Permo-Carboniferous and Triassic magmatic underplating in the IVZ (ZINGG, 1983; VON QUADT et al., 1993; FRANZ et al., 1994; HENK et al., 1997; HANDY et al., in press). Similar to the IVZ, the SCZ only experienced an Alpine sub-greenschist facies overprint (ZINGG, 1990).

Despite these abundant textural, petrologic, and geochronological constraints that indicate a pre-Variscian age for D1 and D2, the metamorphic evolution of the SCZ is a matter of controversy. D3 in the SCZ is related to the amalgamation of the Ivrea-Verbano segment with the Strona-Ceneri segment (HANDY et al., in press). D2 and D1, however, are by some groups interpreted as early phases of docking of IVZ and SCZ and thus of Permo-Carboniferous age (e.g., SCHMID, 1993; BORIANI and VILLA, 1997), whereas ZURBRIGGEN et al. (1997) and other groups argue that these structures originate from a distinctly older orogenic event.

#### Petrologic characterization of the D2 event

Strong reworking of large parts of the SCZ during the D3 tectonometamorphic event, local reheating during the Upper Carboniferous magmatic underplating from the IVZ, and local contact metamorphism near the Permian Baveno-type granites, give the SCZ a complex polydeformative and polymetamorphic character that makes it difficult to find samples suited to date unequivocally pre-D3 metamorphism. However, in the area between Lago d'Orta and Lago Maggiore east of the city of Omegna (Fig. 1), these effects are absent or only weakly developed. There occur monotonous successions of metapelites (mainly garnet-staurolite mica schists) and interlayering garnet-bearing gneisses of psammitic origin. Metasediments in this region mainly carry the textural features of the D2-deformation, i.e., isoclinal folds, with a penetrative S2 schistosity, a uniformly N–S directed stretching lineation, and rotated garnets and syntectonic staurolite, mica, as well as plagioclase.

Five garnet mica schists along a profile from Armeno in the NW to Stresa near the Lago Maggiore (see Fig. 1) were selected and investigated with the microprobe. Two samples are garnet-chlorite mica schists (GC-Ms) which show an alignment of the mineral assemblage Grt + Bt +

Tab. 1 Microprobe analyses of garnet, biotite and staurolite.

| Garnet                         | IZ43   |        | IZ40   |        | IZ50   |        | IZ50   |        | IZ49  |       | IZ46  |       | Biotite | IZ43                           |       | IZ40  |       | IZ50  |       | IZ49  |                                | IZ46   |        | Staurolite | IZ43   |                                | IZ40   |        | IZ50   |        |       |       |  |  |
|--------------------------------|--------|--------|--------|--------|--------|--------|--------|--------|-------|-------|-------|-------|---------|--------------------------------|-------|-------|-------|-------|-------|-------|--------------------------------|--------|--------|------------|--------|--------------------------------|--------|--------|--------|--------|-------|-------|--|--|
|                                | GS-Ms  | GS-Ms  | GS-Ms  | GS-Ms  | GS-Ms  | GS-Ms  | GS-Ms  | GS-Ms  | GS-Ms | GS-Ms | GS-Ms | GC-Ms |         | wt.-%                          | GS-Ms | GS-Ms | GS-Ms | GS-Ms | GS-Ms | GS-Ms | GS-Ms                          | GS-Ms  | GC-Ms  |            | GC-Ms  | wt.-%                          | GS-Ms  | GS-Ms  | GS-Ms  | GS-Ms  | GS-Ms | GS-Ms |  |  |
| wt.-%                          |        |        |        |        |        |        |        |        |       |       |       |       | wt.-%   |                                |       |       |       |       |       |       |                                |        |        | wt.-%      |        |                                |        |        |        |        |       |       |  |  |
| SiO <sub>2</sub>               | 37.32  | 37.44  | 37.36  | 36.84  | 37.34  | 37.22  | 37.59  | 36.52  | 36.53 | 36.22 | 35.42 | 36.36 | 35.40   | SiO <sub>2</sub>               | 36.53 | 36.22 | 35.42 | 36.36 | 36.64 | 35.40 | SiO <sub>2</sub>               | 28.71  | 27.71  | 28.97      | 28.67  | SiO <sub>2</sub>               | 28.71  | 27.71  | 28.97  | 28.67  |       |       |  |  |
| TiO <sub>2</sub>               | 0.09   | 0.05   | 0.04   | 0.10   | 0.00   | 0.01   | 0.05   | 0.12   | 1.81  | 1.67  | 1.71  | 1.45  | 1.50    | TiO <sub>2</sub>               | 1.81  | 1.67  | 1.71  | 1.45  | 1.61  | 1.50  | TiO <sub>2</sub>               | 0.66   | 0.63   | 0.61       | 0.75   | TiO <sub>2</sub>               | 0.66   | 0.63   | 0.61   | 0.75   |       |       |  |  |
| Al <sub>2</sub> O <sub>3</sub> | 21.42  | 21.31  | 21.25  | 21.43  | 21.50  | 21.19  | 20.62  | 21.13  | 18.60 | 19.01 | 19.69 | 18.10 | 18.77   | Al <sub>2</sub> O <sub>3</sub> | 18.60 | 19.01 | 19.69 | 18.10 | 18.95 | 18.77 | Al <sub>2</sub> O <sub>3</sub> | 54.53  | 54.47  | 53.56      | 53.67  | Al <sub>2</sub> O <sub>3</sub> | 54.53  | 54.47  | 53.56  | 53.67  |       |       |  |  |
| Cr <sub>2</sub> O <sub>3</sub> | 0.03   | 0.04   | 0.00   | 0.03   | 0.03   | 0.02   | 0.04   | 0.02   | 0.00  | 0.04  | 0.00  | 0.00  | 0.00    | Cr <sub>2</sub> O <sub>3</sub> | 0.00  | 0.04  | 0.00  | 0.00  | 0.00  | 0.00  | Cr <sub>2</sub> O <sub>3</sub> | 0.07   | 0.09   | 0.00       | 0.03   | Cr <sub>2</sub> O <sub>3</sub> | 0.07   | 0.09   | 0.00   | 0.03   |       |       |  |  |
| Fe <sub>2</sub> O <sub>3</sub> | 0.45   | 0.00   | 0.48   | 0.42   | 0.17   | 0.00   | 0.67   | 1.00   | 10.68 | 10.60 | 10.52 | 11.43 | 11.36   | MgO                            | 10.68 | 10.60 | 10.52 | 11.43 | 10.15 | 11.36 | MgO                            | 1.52   | 1.41   | 1.95       | 1.57   | MgO                            | 1.52   | 1.41   | 1.95   | 1.57   |       |       |  |  |
| MgO                            | 3.40   | 3.23   | 3.26   | 2.26   | 3.33   | 3.09   | 2.72   | 3.25   | 0.00  | 0.00  | 0.00  | 0.00  | 0.00    | CaO                            | 0.00  | 0.00  | 0.00  | 0.00  | 0.00  | 0.00  | CaO                            | 0.00   | 0.00   | 0.01       | 0.00   | CaO                            | 0.00   | 0.00   | 0.01   | 0.00   |       |       |  |  |
| CaO                            | 2.12   | 2.24   | 2.31   | 5.12   | 2.82   | 0.92   | 3.48   | 2.79   | 0.09  | 0.00  | 0.07  | 0.15  | 0.06    | MnO                            | 0.09  | 0.00  | 0.07  | 0.15  | 0.08  | 0.06  | MnO                            | 0.25   | 0.38   | 0.18       | 0.43   | MnO                            | 0.25   | 0.38   | 0.18   | 0.43   |       |       |  |  |
| MnO                            | 1.38   | 1.71   | 2.70   | 2.71   | 1.11   | 1.67   | 2.35   | 2.36   | 19.04 | 18.90 | 18.12 | 18.45 | 18.50   | FeO                            | 19.04 | 18.90 | 18.12 | 18.45 | 18.68 | 18.50 | FeO                            | 12.29  | 11.99  | 13.14      | 12.79  | FeO                            | 12.29  | 11.99  | 13.14  | 12.79  |       |       |  |  |
| FeO                            | 35.09  | 33.96  | 33.60  | 31.61  | 34.46  | 36.11  | 32.92  | 33.05  | 0.00  | 0.00  | 0.10  | 0.05  | 0.06    | CuO                            | 0.00  | 0.00  | 0.10  | 0.05  | 0.00  | 0.06  | CuO                            | 0.00   | 0.00   | 0.00       | 0.00   | CuO                            | 0.00   | 0.00   | 0.00   | 0.00   |       |       |  |  |
| Na <sub>2</sub> O              | 0.01   | 0.03   | 0.00   | 0.01   | 0.00   | 0.02   | 0.03   | 0.01   | 0.15  | 0.26  | 0.22  | 0.13  | 0.11    | Na <sub>2</sub> O              | 0.15  | 0.26  | 0.22  | 0.13  | 0.11  | 0.11  | Na <sub>2</sub> O              | 0.50   | 1.94   | 0.42       | 0.50   | Na <sub>2</sub> O              | 0.50   | 1.94   | 0.42   | 0.50   |       |       |  |  |
| Total:                         | 101.30 | 100.01 | 100.99 | 100.52 | 100.77 | 100.23 | 100.46 | 100.26 | 8.84  | 8.65  | 8.87  | 9.22  | 9.10    | K <sub>2</sub> O               | 8.84  | 8.65  | 8.87  | 9.22  | 8.84  | 9.10  | K <sub>2</sub> O               | 0.00   | 0.00   | 0.00       | 0.00   | K <sub>2</sub> O               | 0.00   | 0.00   | 0.00   | 0.00   |       |       |  |  |
| Cations (O=12)                 |        |        |        |        |        |        |        |        | 3.78  | 3.789 | 3.68  | 3.72  | 3.68    | Na <sub>2</sub> O              | 3.78  | 3.789 | 3.68  | 3.72  | 3.77  | 3.68  | Na <sub>2</sub> O              | 0.01   | 0.06   | 0.03       | 0.00   | Na <sub>2</sub> O              | 0.01   | 0.06   | 0.03   | 0.00   |       |       |  |  |
| Si                             | 2.965  | 3.000  | 2.977  | 2.951  | 2.974  | 2.995  | 3.011  | 2.937  | 0.39  | 0.38  | 0.54  | 0.48  | 0.54    | F                              | 0.39  | 0.38  | 0.54  | 0.48  | 0.40  | 0.54  | F                              | 0.00   | 0.02   | 0.01       | 0.01   | F                              | 0.00   | 0.02   | 0.01   | 0.01   |       |       |  |  |
| Ti                             | 0.005  | 0.003  | 0.002  | 0.006  | 0.000  | 0.001  | 0.003  | 0.007  | 0.08  | 0.04  | 0.08  | 0.06  | 0.03    | Cl                             | 0.08  | 0.04  | 0.08  | 0.06  | 0.05  | 0.03  | Cl                             | 0.00   | 0.00   | 0.00       | 0.00   | Cl                             | 0.00   | 0.00   | 0.00   | 0.00   |       |       |  |  |
| Al                             | 2.006  | 2.013  | 1.996  | 2.023  | 2.018  | 2.009  | 1.946  | 2.003  | 99.99 | 99.56 | 99.01 | 99.60 | 99.10   | Total:                         | 99.99 | 99.56 | 99.01 | 99.60 | 99.28 | 99.10 | Total:                         | 98.53  | 98.68  | 98.89      | 98.00  | Total:                         | 98.53  | 98.68  | 98.89  | 98.00  |       |       |  |  |
| Cr                             | 0.002  | 0.002  | 0.000  | 0.002  | 0.002  | 0.001  | 0.002  | 0.001  | 99.81 | 99.39 | 98.77 | 99.38 | 98.87   | Total c.:                      | 99.81 | 99.39 | 98.77 | 99.38 | 99.10 | 98.87 | Total c.:                      | 3.930  | 3.824  | 3.967      | 3.945  | Total c.:                      | 3.930  | 3.824  | 3.967  | 3.945  |       |       |  |  |
| Fe <sup>3+</sup>               | 0.027  | 0.000  | 0.029  | 0.025  | 0.010  | 0.000  | 0.041  | 0.061  | 2.745 | 2.729 | 2.687 | 2.749 | 2.692   | Si                             | 2.745 | 2.729 | 2.687 | 2.749 | 2.765 | 2.692 | Si                             | 0.068  | 0.065  | 0.063      | 0.077  | Si                             | 0.068  | 0.065  | 0.063  | 0.077  |       |       |  |  |
| Mg                             | 0.403  | 0.386  | 0.387  | 0.270  | 0.395  | 0.370  | 0.325  | 0.390  | 0.103 | 0.095 | 0.098 | 0.082 | 0.086   | Ti                             | 0.103 | 0.095 | 0.098 | 0.082 | 0.091 | 0.086 | Ti                             | 8.797  | 8.857  | 8.644      | 8.703  | Ti                             | 8.797  | 8.857  | 8.644  | 8.703  |       |       |  |  |
| Ca                             | 0.181  | 0.193  | 0.197  | 0.439  | 0.241  | 0.079  | 0.299  | 0.240  | 1.647 | 1.688 | 1.760 | 1.613 | 1.683   | Al                             | 1.647 | 1.688 | 1.760 | 1.613 | 1.686 | 1.683 | Al                             | 0.007  | 0.009  | 0.000      | 0.003  | Al                             | 0.007  | 0.009  | 0.000  | 0.003  |       |       |  |  |
| Mn                             | 0.093  | 0.116  | 0.182  | 0.184  | 0.075  | 0.114  | 0.159  | 0.161  | 0.000 | 0.002 | 0.000 | 0.000 | 0.000   | Cr                             | 0.000 | 0.002 | 0.000 | 0.000 | 0.000 | 0.000 | Cr                             | 0.310  | 0.290  | 0.399      | 0.321  | Cr                             | 0.310  | 0.290  | 0.399  | 0.321  |       |       |  |  |
| Fe <sup>2+</sup>               | 2.331  | 2.276  | 2.239  | 2.117  | 2.295  | 2.430  | 2.205  | 2.223  | 1.196 | 1.191 | 1.189 | 1.288 | 1.288   | Mg                             | 1.196 | 1.191 | 1.189 | 1.288 | 1.142 | 1.288 | Mg                             | 0.000  | 0.000  | 0.000      | 0.000  | Mg                             | 0.000  | 0.000  | 0.000  | 0.000  |       |       |  |  |
| Na                             | 0.001  | 0.005  | 0.000  | 0.002  | 0.000  | 0.004  | 0.004  | 0.002  | 0.000 | 0.000 | 0.000 | 0.000 | 0.000   | Ca                             | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | Ca                             | 1.407  | 1.383  | 1.505      | 1.472  | Ca                             | 1.407  | 1.383  | 1.505  | 1.472  |       |       |  |  |
| Total:                         | 8.013  | 7.990  | 8.009  | 8.019  | 8.011  | 8.019  | 7.996  | 8.025  | 0.006 | 0.000 | 0.004 | 0.010 | 0.004   | Mn                             | 0.006 | 0.000 | 0.004 | 0.010 | 0.005 | 0.004 | Mn                             | 0.000  | 0.000  | 0.000      | 0.000  | Mn                             | 0.000  | 0.000  | 0.000  | 0.000  |       |       |  |  |
| Endmembers:                    |        |        |        |        |        |        |        |        | 1.197 | 1.191 | 1.149 | 1.167 | 1.179   | Fe                             | 1.197 | 1.191 | 1.149 | 1.167 | 1.179 | 1.177 | Fe                             | 0.050  | 0.197  | 0.043      | 0.051  | Fe                             | 0.050  | 0.197  | 0.043  | 0.051  |       |       |  |  |
| Uvar:                          | 0.09   | 0.12   | 0.00   | 0.09   | 0.10   | 0.05   | 0.12   | 0.07   | 0.000 | 0.000 | 0.003 | 0.002 | 0.000   | Ba                             | 0.000 | 0.000 | 0.003 | 0.002 | 0.000 | 0.002 | Ba                             | 0.000  | 0.000  | 0.000      | 0.000  | Ba                             | 0.000  | 0.000  | 0.000  | 0.000  |       |       |  |  |
| Adr:                           | 0.26   | 0.15   | 0.39   | 0.29   | 0.11   | 0.04   | 2.18   | 0.39   | 0.021 | 0.038 | 0.032 | 0.019 | 0.016   | Na                             | 0.021 | 0.038 | 0.032 | 0.019 | 0.016 | 0.016 | Na                             | 0.002  | 0.015  | 0.008      | 0.000  | Na                             | 0.002  | 0.015  | 0.008  | 0.000  |       |       |  |  |
| Grs:                           | 5.67   | 6.21   | 6.17   | 14.22  | 7.80   | 2.55   | 7.70   | 7.51   | 0.847 | 0.831 | 0.858 | 0.889 | 0.883   | K                              | 0.847 | 0.831 | 0.858 | 0.889 | 0.851 | 0.883 | K                              | 0.000  | 0.003  | 0.001      | 0.002  | K                              | 0.000  | 0.003  | 0.001  | 0.002  |       |       |  |  |
| Alm:                           | 77.52  | 76.61  | 74.50  | 70.33  | 76.34  | 81.19  | 73.80  | 73.76  | 7.763 | 7.766 | 7.780 | 7.817 | 7.830   | Total:                         | 7.763 | 7.766 | 7.780 | 7.817 | 7.735 | 7.830 | Total:                         | 14.601 | 14.687 | 14.653     | 14.625 | Total:                         | 14.601 | 14.687 | 14.653 | 14.625 |       |       |  |  |
| Sps:                           | 3.08   | 3.91   | 6.06   | 6.11   | 2.50   | 3.80   | 5.33   | 5.33   | 1.897 | 1.904 | 1.860 | 1.877 | 1.867   | OH                             | 1.897 | 1.904 | 1.860 | 1.877 | 1.898 | 1.867 | OH                             | 0.000  | 0.000  | 0.000      | 0.000  | OH                             | 0.000  | 0.000  | 0.000  | 0.000  |       |       |  |  |
| Prp:                           | 13.39  | 12.99  | 12.88  | 8.98   | 13.15  | 12.37  | 10.86  | 12.95  | 0.093 | 0.091 | 0.130 | 0.115 | 0.095   | F                              | 0.093 | 0.091 | 0.130 | 0.115 | 0.095 | 0.129 | F                              | 0.000  | 0.000  | 0.000      | 0.000  | F                              | 0.000  | 0.000  | 0.000  | 0.000  |       |       |  |  |
|                                |        |        |        |        |        |        |        |        | 0.010 | 0.005 | 0.010 | 0.008 | 0.006   | Cl                             | 0.010 | 0.005 | 0.010 | 0.008 | 0.006 | 0.004 | Cl                             | 0.000  | 0.000  | 0.000      | 0.000  | Cl                             | 0.000  | 0.000  | 0.000  | 0.000  |       |       |  |  |



Ms + Chl + Pl ± Rt, Ilm within the S2 schistosity. Three samples are garnet-staurolite mica schists (GS-Ms) with the syntectonic D2 mineral assemblage Grt + St + Bt + Ms + Chl + Pl ± Rt, Ilm.

Garnet from the GC-Ms displays compositional growth zoning with bell shaped patterns for Mn and Ca and progressively increasing Fe- and Mg-contents from core to rim. In near rim sections, garnet yields the highest  $X_{Mg}$  (= Mg/(Mg + Fe)) values (0.86–0.88) and shows a composition of  $Alm_{73-14}Prp_{11-13}Grs_{7-8}Sps_{5-6}$ . Plagioclase is also often zoned with core compositions of  $An_{04-06}$  steadily increasing to  $An_{18-19}$  at the rim, whereas sheet silicates are unzoned (see Tab. 1 and 2 for mineral compositions).

Two generations of garnet are present in most GS-Ms: large garnets (> 1 cm) have bell-shaped Mn- and Ca-zonation patterns similar to the garnet from GC-Ms and display the highest  $X_{Mg}$ -values (0.85–0.86) near the rim (see Tab. 1). At the outermost part of the rim, a decrease of the pyrope content at the expense of almandine and spessartine components points to retrograde

cooling. The second generation of garnet occurs as small, idioblastic inclusions with diameters of 0.1–0.2 mm in large, aligned plagioclase grains. These small garnets yield similar  $X_{Mg}$ -values and compositions as the rims of the large garnet grains. They are believed to have grown syntectonically near the peak of D2 metamorphism. Prismatic staurolite is always distinctly aligned in the S2 foliation (Fig. 2) and shows increasing  $X_{Fe}$  (= Fe/(Fe + Mg)) values from core to rim (see Tab. 1). Plagioclase in the GS-Ms is either unzoned or shows a slight decrease in An from core to rim (i.e.,  $An_{22}$ – $An_{16}$ ). This zonation is present in plagioclase inclusions in the bigger garnets. Mineral compositions of plagioclase as well as the sheet silicates of the GS-Ms are listed in tables 1 and 2.

Peak metamorphic temperatures for the D2 event in GC-Ms and GS-Ms, estimated with the Grt-Bt-geothermometer of HODGES and SPEAR (1982), the Grt-St-geothermometer of PERCHUK (1969) and the Grt-Chl-geothermometer of GHENT et al. (1987), were in the range of 570–610 °C. Metamorphic pressures were at 7.4–8.4 kbar (Fig. 1) according to the Grt-Ms-Bt-Plg-geobarometer of HOISCH (1989; calibration P1) and is in accordance with the calibration of POWELL and HOLLAND (1988) and the GRIPS geobarometer of BOHLEN and LIOTTA (1986). Sample IZ50 allows a determination of a clockwise, prograde P-T-t path (see Fig. 3) using plagioclase inclusions in the garnet core ( $An_{21}$ ), in the near rim section of the garnet ( $An_{19}$ ) and a plagioclase rim ( $An_{16}$ ) adjacent to the retrograde rim section of garnet in

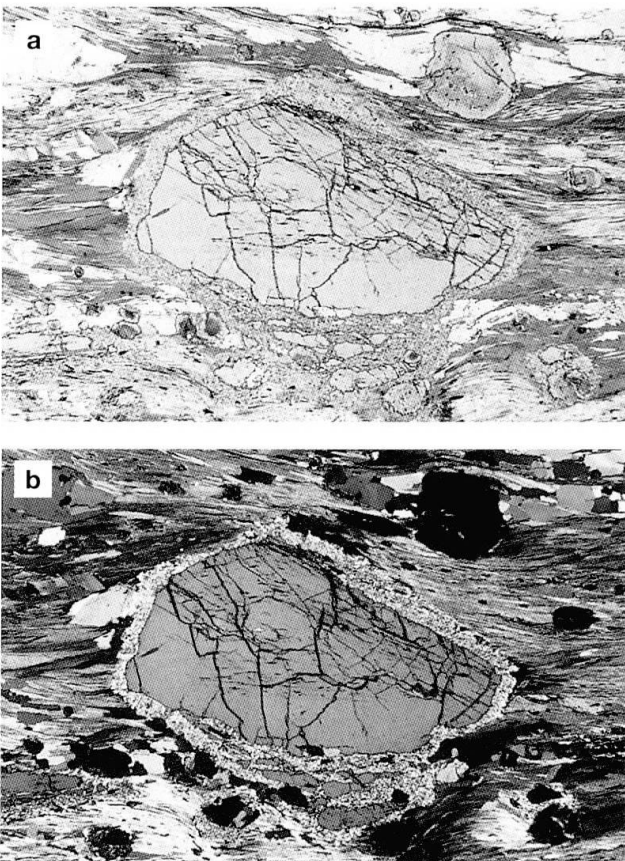


Fig. 2 Microphotograph of syntectonic staurolite with inclusion trails of ilmenite, rutile, and graphite aligned in the D2 foliation along with garnet, biotite, muscovite, plagioclase, and quartz. Photograph with crossed nicols distinctly shows decomposition of the staurolite to chlorite and mica at its rim. Length of lower edge is 7 mm.

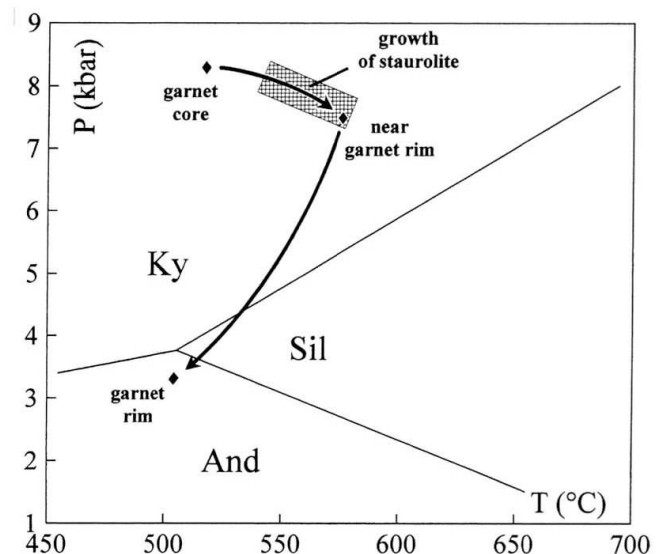


Fig. 3 P-T path for the D2 event derived by garnet-biotite thermometry of HODGES and SPEAR (1982) and garnet-biotite-muscovite-plagioclase geobarometry of HOISCH (1989).

Tab. 3 U–Pb analytical data of staurolite from a garnet-staurolite mica schist from the Strona-Ceneri Zone, east of Omegna, northern Italy.

| Sample <sup>a</sup>                                                                                                                                                   | Weight (mg) | Concentrations (ppm) |                   | Radiogenic Pb (at%) <sup>c</sup>                                  |                   |                   |                   | Atomic ratios <sup>c</sup>          |                                     |                                      | Apparent ages (Ma) <sup>d</sup>     |                                     |                                      |
|-----------------------------------------------------------------------------------------------------------------------------------------------------------------------|-------------|----------------------|-------------------|-------------------------------------------------------------------|-------------------|-------------------|-------------------|-------------------------------------|-------------------------------------|--------------------------------------|-------------------------------------|-------------------------------------|--------------------------------------|
|                                                                                                                                                                       |             | U                    | Pb <sub>tot</sub> | <sup>206</sup> Pb/ <sup>204</sup> Pb Measured ratios <sup>b</sup> | <sup>206</sup> Pb | <sup>207</sup> Pb | <sup>208</sup> Pb | <sup>206</sup> Pb/ <sup>238</sup> U | <sup>207</sup> Pb/ <sup>235</sup> U | <sup>207</sup> Pb/ <sup>206</sup> Pb | <sup>206</sup> Pb/ <sup>238</sup> U | <sup>207</sup> Pb/ <sup>235</sup> U | <sup>207</sup> Pb/ <sup>206</sup> Pb |
| Common lead $^{206}\text{Pb}/^{204}\text{Pb} = 18.77 \pm .07$ , $^{207}\text{Pb}/^{204}\text{Pb} = 15.67 \pm .02$ , $^{208}\text{Pb}/^{204}\text{Pb} = 37.77 \pm .07$ |             |                      |                   |                                                                   |                   |                   |                   |                                     |                                     |                                      |                                     |                                     |                                      |
| St1                                                                                                                                                                   | 6.455       | 0.603                | 0.234             | 34.066                                                            | 36.60             | 2.01              | 61.39             | .06150                              | .46576                              | .05493                               | 385                                 | 388                                 | 409                                  |
| St2                                                                                                                                                                   | 3.258       | 3.789                | 1.015             | 44.514                                                            | 37.96             | 2.17              | 59.87             | .05735                              | .45257                              | .05724                               | 360                                 | 379                                 | 501                                  |

<sup>a</sup> Staurolite concentrates were obtained by separating millimeter large grains from the garnet-staurolite-mica schist, crushing these grains and separating the fragments by hand under the binocular. Care was taken to use only fragments with fracture surfaces to avoid the thin films of plagioclase and retrograde chlorite that have formed locally on grain surfaces. All analyzed fragments were optically clear and inclusion-free. Both samples were dissolved with 52% HF in Savillex beakers at 220 °C for six days on the hot plate, dried, and transferred into chloride-form using 6N HCl. Ion-exchange chromatography, after aliquotting by weight and tracer-addition, as described by TILTON (1973) and MANHÈS et al. (1978). Measured on single Re-filaments using a silica-gel emitter and H<sub>3</sub>PO<sub>4</sub> (GERSTENBERGER and HAASE, 1997) at 1200–1260 °C on a Finnigan MAT262 multicollector mass-spectrometer using Faraday collectors.

<sup>b</sup> Lead isotope ratios corrected for fractionation with 0.1% / a.m.u.

<sup>c</sup> Lead corrected for fractionation, blank, and initial lead. During the measurement period total blanks were less than 30 pg for lead and less than 1 pg for uranium for samples analyzed with a <sup>208</sup>Pb–<sup>235</sup>U mixed tracer.

<sup>d</sup> Apparent ages were calculated using the constants of JAFFEY et al. (1971) recommended by IUGS (STEIGER and JÄGER, 1977).

combination with muscovite and biotite from the matrix. These results agree with the contoured petrogenetic grid of SPEAR and CHENEY (1989), with the geothermometry of POWNCEBY et al. (1987 a, b) on ilmenite inclusions in garnet, and the GRIPS-geobarometer of BOHLEN and LIOTTA (1986).

### U–Pb and Pb–Pb systematics of staurolite

Two staurolite samples have been analyzed using U–Pb (Tab. 3). The distribution of uranium and common lead in staurolite seems to be heterogeneous as indicated by the contents of 0.6 and 3.8 ppm of uranium (Tab. 3) and ca 0.15 and 0.52 ppm of common lead. Because of the common lead content, the isotopic composition of staurolite is not very radiogenic.  $^{206}\text{Pb}/^{204}\text{Pb}$  ratios of the two samples are 34.066 and 44.514 and the corresponding  $^{207}\text{Pb}/^{204}\text{Pb}$  ratios are only 16.508 and 17.081. Consequently, the contributions of initial lead are significant and the  $^{207}\text{Pb}_{\text{rad}}/^{235}\text{U}$  and  $^{207}\text{Pb}_{\text{rad}}/^{206}\text{Pb}_{\text{rad}}$  ratios have rather large errors 2.7–4.7% and 2.6–4.4% respectively. Due to the large uncertainty in the  $^{207}\text{Pb}_{\text{rad}}/^{235}\text{U}$  ratio, the data fall in the concordia diagram (Fig. 4) on the concordia. The contrasting  $^{206}\text{Pb}_{\text{rad}}/^{238}\text{U}$  ratios, however, indicate that the two samples have not been closed systems, but seem to have lost some lead.

Thus, the well-constrained  $^{206}\text{Pb}/^{238}\text{U}$  ages ( $385 \pm 6$  Ma and  $360 \pm 3$  Ma, Tab. 3) represent minimum ages for the crystallization of staurolite. The minimum age for the crystallization of staurolite is given by the higher of the two  $^{206}\text{Pb}/^{238}\text{U}$  ages, i.e.,  $385 \pm 6$  Ma (Tab. 3; Fig. 4). The staurolite minimum age supports K–Ar hornblende ages between 346 Ma and 389 Ma from the northeastern part of the SCZ that have been interpreted to yield minimum ages for a pre-Variscan event (MCDOWELL, 1970). Notably, both staurolite minimum ages and

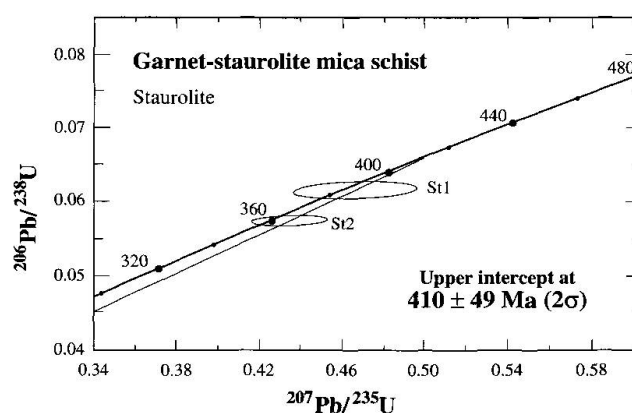


Fig. 4 Concordia diagram for staurolite bulk-samples St1 and St2. The contrasting  $^{206}\text{Pb}/^{238}\text{U}$  age of the two fractions indicates lead loss. The intercept age of  $410 \pm 49$  Ma is calculated from a discordia fitted through the origin of the diagram and represents a minimum age.



the pioneering K–Ar hornblende ages of McDOWELL (1970) are beyond error higher than the  $^{39}\text{Ar}/^{40}\text{Ar}$  ages on amphibole and mica from rocks of the southwestern part of the SCZ (BORIANI and VILLA, 1997).

A third staurolite sample was leached with successively stronger acids and the lead isotopic composition of the leachates was analyzed (Tab. 4). The leaching procedure (Tab. 4) follows the one of FREY and KAMBER (1995). The idea of this leaching procedure is: (1) For grains that were closed systems after crystallization, the lead data should fall on a straight line, whose slope  $^{207}\text{Pb}/^{206}\text{Pb}$  only depends on the age of the mineral. (2) Lead incorporated in the crystal structure, located on fractures and grain surfaces, and formed by radioactive decay of uranium and thorium behave differently during leaching. Lead on surfaces and on structural sites damaged by radioactive decay are more easily leached than lead incorporated into the crystal structure. Lead that is easily leached by weak acids most likely also represents those lead contributions most easily affected by later metamorphic processes and weathering.

The lead isotope data of the various leachates from the staurolite do not fall on a common straight line in the  $^{206}\text{Pb}/^{204}\text{Pb}$ – $^{207}\text{Pb}/^{204}\text{Pb}$  and  $^{206}\text{Pb}/^{204}\text{Pb}$ – $^{208}\text{Pb}/^{204}\text{Pb}$  diagrams, which agrees with the U–Pb data that also indicate some minor post-crystallization disturbance. The two HBr leachates seem to be the most affected solutions (Fig. 5). Leachate HBr-1 represents a low-temperature leach and most likely has mobilized lead from grain surfaces and fractures (cf. also FREY et al., 1995, 1997). It has the least radiogenic lead isotopic composition of the five solutions. It is distinctive by its – in comparison to the  $\text{HNO}_3$ , HCl and HF leachates – anomalously high  $^{208}\text{Pb}/^{204}\text{Pb}$  ratio (43.314; Tab. 4, Fig. 5). Leachate HBr-2 has the most radiogenic lead isotopic composition of all leachates. It falls in the  $^{206}\text{Pb}/^{204}\text{Pb}$ – $^{207}\text{Pb}/^{204}\text{Pb}$  diagram below the regression line through the other samples and shows in the  $^{206}\text{Pb}/^{204}\text{Pb}$ – $^{208}\text{Pb}/^{204}\text{Pb}$  diagram similarly high  $^{208}\text{Pb}/^{206}\text{Pb}$  signatures as leachate HBr-1. The lead isotope signature of leachate HBr-2 is compatible with a two-stage evolution, whereby the second stage had a higher  $\mu$ -value ( $^{238}\text{U}/^{204}\text{Pb}$ ) than the first one, which corresponds to lead loss or uranium gain. The other three leachates fall on linear trends in the  $^{206}\text{Pb}/^{204}\text{Pb}$ – $^{207}\text{Pb}/^{204}\text{Pb}$  and  $^{206}\text{Pb}/^{204}\text{Pb}$ – $^{208}\text{Pb}/^{204}\text{Pb}$  diagrams; which indicates that they behaved chemically coherent, represent structurally bound lead (HF-leachate is the total dissolution of the residue), and originate from structural sites with similar Th/U.

Tab. 4 Staurolite trace lead isotope data from a garnet-staurolite-mica schist east of Omegna, northern Italy.

| Sample <sup>a</sup> | $^{206}\text{Pb}^b$ | $^{207}\text{Pb}^b$ | $^{208}\text{Pb}^b$ |
|---------------------|---------------------|---------------------|---------------------|
|                     | $^{204}\text{Pb}$   | $^{204}\text{Pb}$   | $^{204}\text{Pb}$   |
| St1                 | 34.066              | 16.508              | 63.438              |
| St2                 | 44.514              | 17.081              | 78.372              |
| St3-HBr-1           | 20.417              | 15.675              | 43.314              |
| St3-HBr-2           | 82.172              | 18.881              | 152.534             |
| St3- $\text{HNO}_3$ | 23.213              | 15.862              | 38.609              |
| St3-HCl             | 20.687              | 15.741              | 38.229              |
| St3-HF              | 27.077              | 16.121              | 39.599              |

- <sup>a</sup> Bulk staurolite samples St1 and St2. Sample St3 was ground and washed with 0.8 N HBr before leaching with cold 8.8 N HBr for 3 h (in ultrasonic). Other leaching steps were performed on the hot plate: with 8.8 N HBr for 24 h, with 14 N  $\text{HNO}_3$  for 24 h, with 6 N HCl for 24 h, and with 52% HF for 72 h.
- <sup>b</sup> Lead isotope analyses were performed at GeoForschungsZentrum Potsdam, Germany, using a Finnigan MAT262 multicollector mass spectrometer. The lead isotopic composition is corrected for mass discrimination with 0.1%/a.m.u.  $2\sigma$  reproducibility is better than 0.1%.

The contrasting  $^{208}\text{Pb}_{\text{rad}}/^{206}\text{Pb}_{\text{rad}}$  (measured lead corrected for contributions of common lead as indicated in Tab. 3) signature of the HBr-leachates and the other three leachates indicates that Th and U in part are located in different sites in the staurolite crystal. The higher  $^{208}\text{Pb}_{\text{rad}}/^{206}\text{Pb}_{\text{rad}}$  ratios of the HBr-leachates indicate a Th/U of 13.5 and 7.3 (derived for an age of 450 Ma), whereas the Th/U of the other fractions ranges from 0.76 to 0.96. The easier leached lead, i.e., the lead bound to fractures (cf. Fig. 2) and crystal imperfections, has a higher Th/U than the structurally bound lead. Furthermore, the high Th/U of leachate HBr-2 indicates that most of the thorium is not incorporated in the crystal structure of staurolite, but resides in defects and imperfections. The slope of the lead line corresponds to a  $^{207}\text{Pb}/^{206}\text{Pb}$  age at  $282 \pm 15$  Ma, which falls in the same age range as the post-kinematic Permian intrusions of Baveno, Montorfano, and Mottarone (e.g., PINARELLI et al., 1988). This  $^{207}\text{Pb}/^{206}\text{Pb}$  age is a geologically significant age if (1) the initial common lead of the two HBr-leachates was identical and (2) the geologic disturbance affecting the lead systematics of leachate HBr-2 caused a complete removal of radiogenic lead. Else, the  $^{207}\text{Pb}/^{206}\text{Pb}$  age constrained by these two samples represents a minimum age. In contrast, the  $^{207}\text{Pb}/^{206}\text{Pb}$  age of the lead line through the remaining scattered three samples is 605 Ma (MSWD = 10.5) with a very

large error. Using these three samples and the bulk samples St1 and St2 yields  $477 \pm 120$  Ma ( $2\sigma$ ; MSWD = 6.5; uncertainties expanded with square root of MSWD, see e.g., KULLERUD, 1991; LUDWIG, 1994), most of the scatter being due to sample St3-HNO<sub>3</sub> (note, that samples St1 and St2 do not strictly conform with the requirements for a fit to a secondary-lead isochron [e.g., ROMER and BRIDGWATER, 1997] as the bulk-samples

have suffered from Variscan lead loss and may reduce the slope and apparent age of the lead line). Alternatively to the interpretations that the two sets of lead data define secondary lead isochrons, the lead lines also could originate from two-stage lead-growth with highly correlated  $\mu_1$  and  $\mu_2$  values (see SCHIØTTE, 1989; ROMER and BRIDGWATER, 1997). For  $\mu_1 < \mu_2$ , as inferred from the U-Pb data (lead loss or uranium gain), the

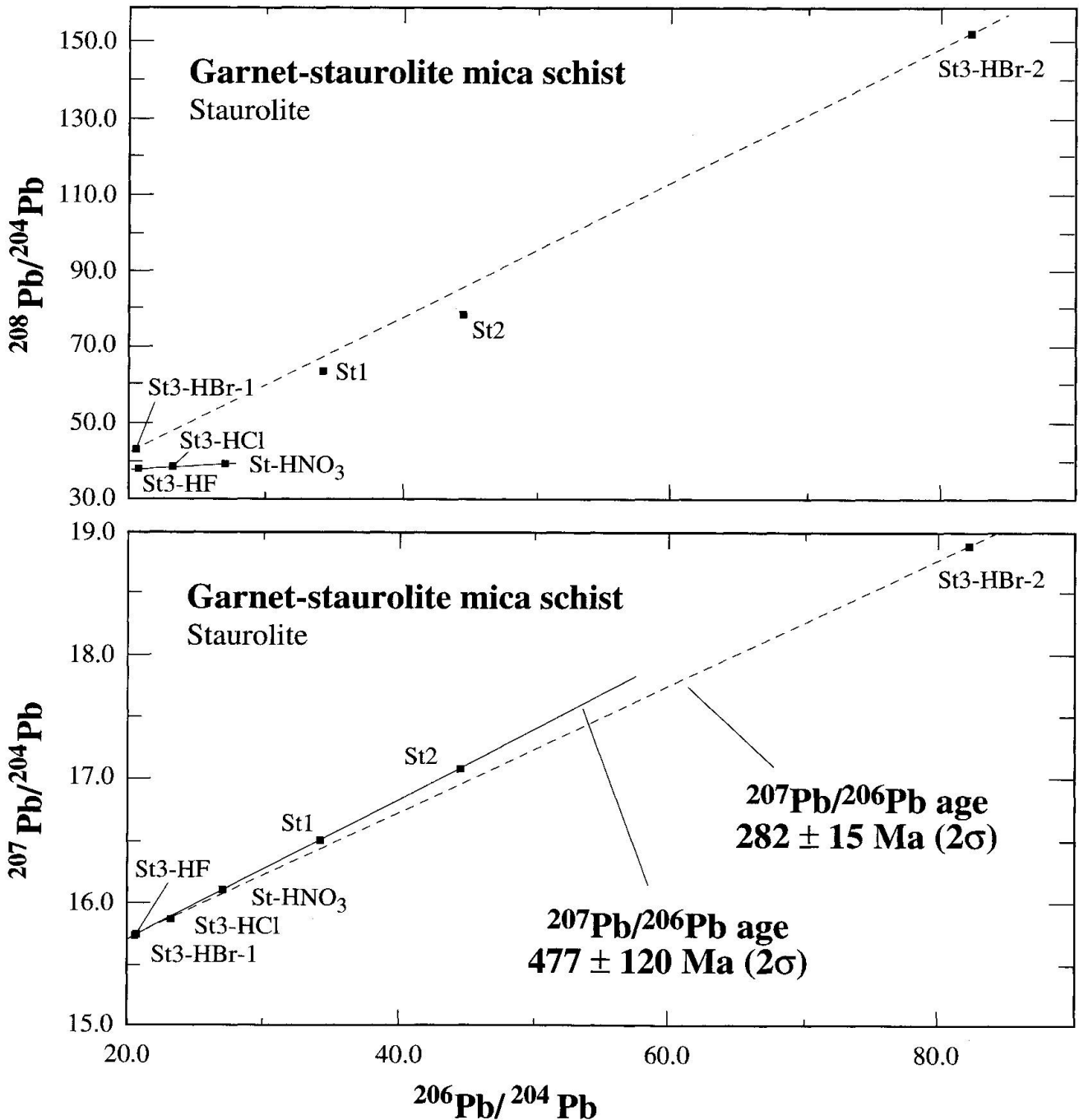


Fig. 5  $^{206}\text{Pb}/^{204}\text{Pb}$ - $^{207}\text{Pb}/^{204}\text{Pb}$  and  $^{206}\text{Pb}/^{204}\text{Pb}$ - $^{208}\text{Pb}/^{204}\text{Pb}$  diagram for staurolite samples. Note (1) the contrasting  $^{207}\text{Pb}/^{206}\text{Pb}$  slopes of the line St3-HBr-1 to St3-HBr-2 and all other fractions and (2) the contrasting  $^{208}\text{Pb}/^{206}\text{Pb}$  slopes for the HBr leachates and the HNO<sub>3</sub>, HCl, and HF leachates and the position of the bulk-samples St1 and St2 between the two lines.

obtained ages at  $282 \pm 15$  and  $477 \pm 120$  Ma would represent minimum ages.

The  $^{206}\text{Pb}/^{204}\text{Pb}$ – $^{208}\text{Pb}/^{204}\text{Pb}$  diagram demonstrates that the lead isotopic signature of the analyzed staurolite is dominated by two components. This is also apparent by the position of the two bulk fractions in that diagram. The samples St1 and St2 represent a weighted mixture of the different leachates and fall correspondingly also between the lead trend of the HBr-leachates and the other leachates (Fig. 5). Furthermore, the presence of two contrasting lead signatures in staurolite also explains the contrasting  $^{206}\text{Pb}/^{238}\text{U}$  ages of the samples St1 and St2 that apparently represent mixed ages between the formation age given by the  $\text{HNO}_3$ – $\text{HCl}$ – $\text{HF}$  lead line (ca 477 Ma) and the HBr-1–HBr-2 lead line (ca 282 Ma). The lead systematics of the leached staurolite sample demonstrates unequivocally the presence of two lead components of contrasting age, which implies that the  $^{206}\text{Pb}/^{238}\text{U}$  age at  $385 \pm 6$  Ma of St1 represents a minimum age for the S2-defining metamorphism, and thus, gives proof for a pre-Variscan metamorphism in the SCZ.

#### **Geochronologic, petrologic, and structural arguments for and against a Variscan age of the D2 deformation**

The oldest metamorphic rocks in the SCZ are D1-deformed amphibolite-facies metapelites and gneisses, as well as amphibolites with eclogitic relics, which all occur as xenoliths within the Ordovician granitoids. The xenoliths have the same age or are older than the granitoids, which intruded ca. 450–460 Ma ago (KÖPPEL and GRÜNENFELDER, 1971; RAGETTLI, 1993). The D2 foliation at the margin and interior of these intrusive rocks demonstrates that the granitoids were emplaced syntectonic with the formation of this foliation (ZURBRIGGEN et al., 1997). Therefore, the age for D2 structures is given by the emplacement age of the oldest intrusive rocks, i.e., the ca. 450–460 Ma old granitoids (KÖPPEL and GRÜNENFELDER, 1971; RAGETTLI, 1993). The D3 event is dated by Pb–Pb age at 320 Ma on garnet from a synkinematic (D3) dike in the Valle Cannobina (ZURBRIGGEN et al., 1997). Furthermore, the correlation of amphibolite and greenschist-facies structures (D3) in the SCZ with  $^{39}\text{Ar}$ – $^{40}\text{Ar}$  and Rb–Sr mica and amphibole ages, which close in the range of 300–500 °C indicate that D3 is of Variscan age. Finally, mylonites in the Val Colla Zone overprint D3 structures. The mylonites are dated and constrain the age of D3 to  $> 320$  Ma (cf. HANDY et al., in press).

These data should be argument enough to assign deformation phase D2 (D1 in BORIANI and VILLA, 1997) to a pre-Variscan orogenic event that is entirely unrelated with the docking of the SCZ with the IVZ. However, because of the geographically widespread mica and amphibole ages (K–Ar,  $^{39}\text{Ar}$ – $^{40}\text{Ar}$ , Rb–Sr) from all kinds of structures in the SCZ (e.g., ZINGG et al., 1990; BORIANI and VILLA, 1997 and references therein) and the common problem to relate the zircon and monazite ages to structures and metamorphic mineral assemblages, BORIANI and VILLA (1997) disqualified all geochronologic work yielding pre-Variscan ages. They argue that these ages are irrelevant to the determination of the age of D2 deformation and metamorphism or flawed by analytical problems. But, their arguments are largely selective and partially misquoted:

(1) Ordovician ages for monazite from Casletto (KÖPPEL and GRÜNENFELDER, 1971) are interpreted by BORIANI and VILLA (1997, p. 386, left middle) to date the contact metamorphism around Ordovician intrusions rather than regional metamorphism. However, there is no evidence for contact metamorphism around these intrusions (see ZINGG, 1983; HANDY, 1986; ZURBRIGGEN et al., 1997).

(2) BORIANI and VILLA (1997, p. 386, right bottom) argue that U–Xe ages (RAGETTLI, 1993) are hampered by uncertainties for the true value of the fission constant and the small analytical reproducibility (scatter among different zircon and monazite fractions from the same sample), and that therefore the U–Xe ages are irrelevant. However, BORIANI and VILLA (1997) do not mention that aliquots of the zircon and monazite samples used for Xe–U dating also have been dated with the U–Pb method. All these monazite and zircon samples yield Ordovician U–Pb ages (RAGETTLI, 1993), i.e.,  $457 \pm 4$  Ma (Giumello gneiss),  $456 \pm 4$  Ma (Ceneri gneiss), and  $452 \pm 3$  Ma (paragneiss from Ruspesso). Actually, combined U–Pb and U–Xe data were used to estimate the fission constant since it had become obvious that the value estimated from uranium oxides, which are potentially susceptible to loss of fissionogenic Xe, and the one derived from chemically more stable silicate and phosphate minerals differ (cf. EIKENBERG et al., 1993; RAGETTLI, 1993; RAGETTLI et al., 1994).

(3) Monazite and zircon in dry systems have a high retentivity for lead and may therefore remain closed systems in amphibolite facies and lower grade metamorphism (e.g., MEZGER and KROGHSTAD, 1997; DAHL, 1997). In the presence of fluids, the U–Pb system of these minerals may behave as an open or partially open system, as these minerals can recrystallize (e.g., LEE, 1993;

TEUFEL and HEINRICH, 1997). BORIANI and VILLA (1997) argue that zircon and monazite ages for some samples, because of the high blocking temperature of these minerals, are derived from inherited components. Other zircon ages originally interpreted to reflect Ordovician high-grade metamorphism (KÖPPEL and GRÜNENFELDER, 1971) are reinterpreted by BORIANI and VILLA (1997) to date diagenesis, whereas the high-grade metamorphism would have gone by unrecorded by the U–Pb system of these zircon samples (BORIANI and VILLA, 1997, p. 386, right top).

The new staurolite data helps to resolve this controversy as staurolite represents (1) a major constituent of S2 structures (as used by ZURBRIGGEN et al., 1997), (2) has a blocking temperature for the U–Pb system that is higher than the thermal maximum experienced by its host rock (see DAHL, 1997), and (3) has no inherited components. The Pb systematics proves an Ordovician age for the growth of staurolite and a Variscan age for partial lead loss along presumably structurally controlled sites. The Variscan lead loss from staurolite implies that the  $^{206}\text{Pb}/^{238}\text{U}$  ages at  $385 \pm 6$  Ma and  $360 \pm 3$  Ma are minimum ages for staurolite crystallization and minimum ages for the metamorphic event related with S2 structures in the SCZ. The best age estimate for this pre-Variscan metamorphism is given by published U–Pb zircon and monazite ages ( $457 \pm 4$  Ma,  $456 \pm 4$  Ma, RAGETTLI [1993] and  $450 \pm 10$ , KÖPPEL and GRÜNENFELDER [1971]). The Permo-Carboniferous minimum age given from the readily leached staurolite-lead demonstrates a second metamorphic event, which agrees with the large number of K–Ar,  $^{39}\text{Ar}$ – $^{40}\text{Ar}$ , and Rb–Sr mica and amphibole ages (for a compilation see HUNZIKER et al., 1992). Therefore, the metamorphic and structural memory of the SCZ was acquired during at least two unrelated orogenic events and refutes models that rule out a pre-Variscan structural and metamorphic history of the SCZ.

### Conclusions

Staurolite is an essential phase of the mineral assemblage that formed during the metamorphism related with deformation D2, which is the major metamorphic event in the sampled part of the SCZ. The U–Pb systematics of staurolite clearly demonstrates that this tectonometamorphic event occurred during the Ordovician. Therefore, the Variscan  $^{39}\text{Ar}$ – $^{40}\text{Ar}$  ages on mica and amphibole are associated with the S3 (S2 in BORIANI and VILLA, 1997) event, having crystallized or having been thermally overprinted and reset during this event.

Furthermore, BORIANI and VILLA's (1997) assertions of zircon recrystallization during Ordovician "diagenesis" (p. 386), apparently an opinion held by a minority of one, is not compatible with the staurolite age data. Similarly, their reinterpretation and affirmation that the ages obtained by the U–Xe method indicate a Variscan event rather than an Ordovician event, lacks support. Instead, the staurolite age data confirm and support earlier work demonstrating Ordovician metamorphism in the SCZ (e.g., PIDGEON et al., 1970; KÖPPEL and GRÜNENFELDER, 1971, 1978; KÖPPEL, 1974; RAGETTLI, 1993).

### Acknowledgement

We thank M. Boche and J. Müller for separation of the staurolite. We gratefully acknowledge discussion and comments by M. Handy (Giessen) and constructive and detailed reviews by V. Köppel and U. Schaltegger (both Zürich).

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Manuscript received February 15, 1998; revision accepted June 22, 1998.